

XXXVII Cycle

Electrochmical devices for carbon capture and valorization Alessio Mezza Supervisor: Prof. Fabrizio Pirri

Research context and motivation

- The problem of global warming, undoubtedly caused by the human activities, becomes more and more pressing on our planet. Large amounts of greenhouse gases (GHGs) have been emitted in the atmosphere, contributing to the rise in temperatures. Among the GHGs, carbon dioxide (CO_2) is the one with the largest concentration in the environment and the longest residence time, thus representing the most impacting climate-altering substance.
- To reduce the CO_2 emissions in the atmosphere, carbon capture and utilization (CCU) technologies have been proposed as effective solutions. In this framework, electrochemical processes demonstrated to be intriguing strategies to implement platforms able to couple capture and valorization, possibly powered by renewable sources allowing a carbon neutral path.



Novel contributions

- The work related to ICCU has been focused on the GDE and on its influence on the performance of zero-gap (fuel cell-like) device: BPM Ag GDL Ag Electrolyte
 - GDL (thickness and hydrophilicity)





• Among the different products obtainable from the electrochemical CO_2 reduction reaction (CO_2RR), techno-economic analyses showed that carbon

monoxide (CO) is the one with the highest profitability. Since in water-based electrolytes, HER is a competing reaction, syngas with tunable composition can be obtained.

Addressed research questions/problems

Coupled systems

The most simple idea is to couple a CO_2 capture module with a conversion module. Several techniques for capture and conversion devices have been deeply investigated in literature. We implemented a platform with an electrohcemical cell for electrodialytic extraction of CO₂ from bicarbonate as capture device, connected with a gas-fed flow cell used as CO_2RR device. The CO_2RR is activated by a nanostructured catalyst of ZnO.



- MEA: nickel foam Fumasep BPM Sigracet carbon paper -- sputtered silver
- Electrolytes: KOH and KHCO₃
- Potentiostat for electrochemical characterizations
- µGC for gasseous products analysis in time
- Physical vapor deposition of silver
- Characterization of catalyst material
 - FESEM



P P



Integrated system

Stripping from capture media is very energy demanding. We focused on the possible integration of the capture and the conversion (ICCU), using the capture media directly as electrolyte for the CO₂RR in fuel cell-like device. The utilization of a MEA, including a bipolar membrane, is fundamental to have local release and conversion into CO of CO_2 .



Submitted and published works

Mezza, A.; Pettigiani, A, and Sacco, A., "An Electrochemical Platform for the Carbon Dioxide Capture and Conversion to Syngas", Energies, vol. 14, no. 23, 2021, pp. 7869

XRD • COMSOL for MEA modeling

Future work

- Increase currents and FE_{CO} improving the morphology of catalyst and GDE
- Implement a stable and efficient plant for Integrated Carbon Capture and Utilization
- Introduce cheaper catalyst material for CO₂RR in zero-gap electrolyzer
- Get a reliable model for MEA in CO₂RR and ICCU systems
- Exploit the zero-gap electrolyzer for low energy CO₂ capture from flue gas



- 02UMPRS An Introduction to Climate Change (28/3/22, 4)
- 01DMMKG Impedance spectroscopy for electrochemical processes (10/2/22, 4)
- 01DMZIY Photo-Electro-Catalytic Technologies for a Sustainable Chemical I. (17/6/22, 6)
- 01DUBRU Understanding electrochemical processes using EIS (21/9/22, 4)
- Introduction to COMSOL MULTIPHYSICS (9/5/22, 12h)
- Short Courses on Electrochemistry (28/6/22, 12h)
- 02LWHRV Communication (14/4/22, 1)
- 01SHMRV Entrepreneurial Finance (12/1/22, 1)
- 08IXTRV Project management (11/1/22, 1)
- 01RISRV Public speaking (14/1/22, 1)
- 01SYBRV Research integrity (13/1/22, 1)



Electrical, Electronics and

Communications Engineering